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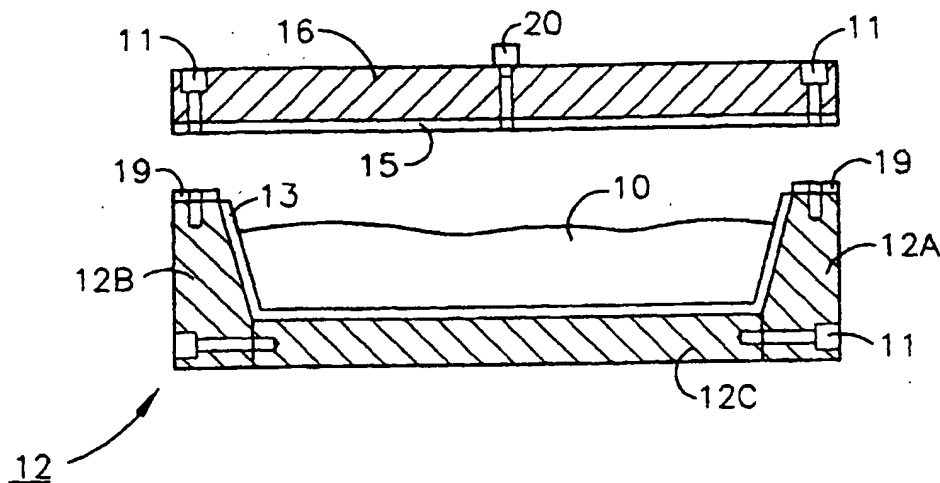
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- (72) Inventor: ROGERS, Darren, Kenneth; 4 America Avenue, Wheeling, WV 26003 (US). For two-letter codes and other abbreviations, refer to the "Guidance Notes on Codes and Abbreviations" appearing at the beginning of each regular issue of the PCT Gazette.

(54) Title: CELLULAR COAL PRODUCTS AND PROCESSES



(57) Abstract: A porous coal-based material (10A) is provided having a density of between about 0.1 gram per cubic centimeter and about 0.6 gram per cubic centimeter produced by the controlled heating of small coal particulate (10) in a "mold" (12) and under a non-oxidizing atmosphere. The porous product (10A) thereby produced, preferably as a near net shape, can be machined, adhered and otherwise fabricated to produce a wide variety of low cost, low density products, or used in its preformed shape (10A) as a filter, heat or electrical insulator etc. These said porous products (10A), without further treatment exhibit compressive strengths of up to about 6000 psi. Further treatment by carbonization or graphitization of said porous products (10A) yield products that can be used as electrical or heat conductors. Methods for the production of these said coal-based cellular products (10A).

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Cellular Coal Products and Processes**Field of the Invention**

5 The present invention relates to cellular coal products produced from coal powder and to their methods of production. Products utilizing the coal-based porous products are also described.

Background of the Invention

10

ASTM standards DD5515-97, "Standard Test Method for the Determination of Swelling Properties of Bituminous Coal" and D720-91 "Standard Test Method for Free Swelling Index of Coal" both define conditions for measuring the inherent property of coals to "swell" upon heating in an uncontrolled combustion situation. Hence, the propensity of coal to swell is well known in the prior art. To the best of our knowledge, however, no one has attempted to take advantage of this property of coals to swell by controllably "swelling" a coal product to obtain a highly useful, low density, porous carbon product.

20

Similarly, very sophisticated processes have been developed for the production of cellular foamed carbon products. Such processes often involve the use of blowing agents and the application of very high pressures in the fabrication

process, and many use highly sophisticated starting materials. These materials, while very lightweight and demonstrating superior strength, tend to be relatively costly, either due to the nature of their starting materials and/or the complexity of their fabrication processes.

5

There exists a wide and varied class of requirements for low-density materials in the construction, aerospace, transportation, metal processing and other industries for which low-density materials are constantly being developed. Many of these materials exhibit properties such as fire resistance that make them uniquely suited to their end use application. In many applications, however, the
10 aforementioned relatively high cost, low-density materials cannot be used because the final application will simply not justify their relatively high cost.

Accordingly, it would be most desirable if a relatively low cost, low-density
15 material demonstrating many of the desirable characteristics of the aforementioned products, such as fire resistance, were available.

Objects of the Invention

20 It is therefore an object of the present invention to provide a relatively low cost, low density product that is suited to application in the construction, aerospace, transportation, metal processing and other industries where such properties are desired.

It is another object of the present invention to provide a simple and low cost method for the production of such products.

5

Summary of the Invention

According to the present invention there are provided coal-based cellular or porous products having a density of preferably between about 0.1 g/cm³ and about 10 0.8 g/cm³ that are produced by the controlled heating of coal particulate preferably up to ¼ inch in diameter in a "mold" and under a non-oxidizing atmosphere. The porous product thereby produced, preferably as a net shape or near net shape, can be machined, adhered and otherwise fabricated to produce a wide variety of low cost, low density products, or used in its preformed shape as a filter, heat or 15 electrical insulator etc. Such cellular products, without further treatment and/or the addition of strengthening additives have been shown to exhibit compressive strengths of up to about 4000 psi. Impregnation with appropriate materials or the incorporation of various strength improving additives can further increase the compressive, tensile and other properties of these cellular materials. Further 20 treatment by carbonization or graphitization yields cellular products that can be used as electrical or heat conductors.

Description of the Drawings

Figure 1 is a graph of showing the general relationship between gas evolution
5 and time/temperature at various operating pressures and temperatures for the
process of the present invention.

Figure 2 is a cross-sectional view of a "mold" containing powdered coal prior
to expansion in accordance with the process of the present invention.

10

Figure 3 is a cross-sectional view of the "mold" of Figure 2 subsequent to
expansion of the powdered coal in accordance with the process of the present
invention.

15

Figure 4 is a cross-sectional diagram of an extruder suitable for the
production of coal-based porous products in accordance with the present invention.

Detailed Description

20

According to the present invention, a preformed, low density, i.e., from about
0.1 to about 0.8 g/cm³, and preferably from about 0.1 to about 0.6g/cm³, cellular
product is produced from powdered coal particulate preferably less than about 1/4
inch in diameter by the controlled heating of the powdered coal in a "mold" under a

non-oxidizing atmosphere. The starting material coal may include bitumen, anthracite, or even lignite, or blends of these, but are preferably bituminous, agglomerating coals that have been comminuted to an appropriate particle size, preferably to a fine powder below about -60 to -80 mesh.

5

The cellular coal-based products described herein are semi-crystalline or more accurately turbostratically-ordered and largely isotropic i.e., demonstrating physical properties that are approximately equal in all directions. The cellular coal-based products of the present invention typically exhibit pore sizes on the order of less than 300μ , although pore sizes of up to 500μ are possible within the operating parameters of the process described. The thermal conductivities of the cellular coal-based products are generally less than about $1.0 \text{ W/m}^\circ\text{K}$. Typically, the cellular coal-based products of the present invention demonstrate compressive strengths on the order of from about 2000 to about 6000 psi at densities of from about 0.4 to about 0.5 g/cm^3 .

The production method of the present invention comprises: 1) heating a coal particulate of preferably small i.e., less than about $\frac{1}{4}$ inch particle size in a "mold" and under a non-oxidizing atmosphere at a heat up rate of from about 1 to about 20°C to a temperature of between about 300 and about 700°C ; 2) soaking at a temperature of between about 300 and 700°C for from about 10 minutes up to about 12 hours to form a preform or finished product; and 3) controllably cooling the preform or finished product to a temperature below about 100°C . The non-

oxidizing atmosphere may be provided by the introduction of inert or non-oxidizing gas into the "mold" at a pressure of from about 0 psi, i.e., free flowing gas, up to about 500 psi. The inert gas used may be any of the commonly used inert or non-oxidizing gases such as nitrogen, helium, argon, CO₂, etc.

5

It is generally not desirable that the reaction chamber be vented or leak during the heating and soaking operation. The pressure of the chamber and the increasing volatile content therein tends to retard further volatilization while the cellular product sinters at the indicated elevated temperatures. If the furnace is vented or leaks during soaking, an insufficient amount of volatile matter may be present to permit inter-particle sintering of the coal particles thus resulting in the formation of a sintered powder as opposed to the desired cellular product. Thus, according to a preferred embodiment of the present process, venting or leakage of non-oxidizing gas and generated volatiles is inhibited consistent with the production of an acceptable cellular product.

10

Additional more conventional blowing agents may be added to the particulate prior to expansion to enhance or otherwise modify the pore-forming operation.

15

The term "mold", as used herein is meant to define a mechanism for providing controlled dimensional forming of the expanding coal. Thus, any chamber into which the coal particulate is deposited prior to or during heating and

which, upon the coal powder attaining the appropriate expansion temperature, contains and shapes the expanding porous coal to some predetermined configuration such as: a flat sheet; a curved sheet; a shaped object; a building block; a rod; tube or any other desired solid shape can be considered a "mold" for purposes of the instant invention.

As will be apparent to the skilled artisan familiar with pressurized gas release reactions, as the pressure in the reaction vessel, in this case the mold increases, from 0 psi to 500 psi, as imposed by the non-oxidizing gas, the reaction time will increase and the density of the produced porous coal will increase as the size of the "bubbles" or pores produced in the expanded coal decreases. Similarly, a low soak temperature at, for example about 400°C will result in a larger pore or bubble size and consequently a less dense expanded coal than would be achieved with a soak temperature of about 600°C. Further, the heat-up rate will also affect pore size, a faster heat-up rate resulting in a smaller pore size and consequently a denser expanded coal product than a slow heat-up rate. These phenomenon are, of course, due to the kinetics of the volatile release reactions which are affected, as just described, by the ambient pressure and temperature and the rate at which that temperature is achieved. These process variables can be used to custom produce the expanded coals of the present invention in a wide variety of controlled densities, strengths etc. These results are graphically represented in the Figure where the X axis is gas release, the Y axis is time and the individual curves represent different

pressures of inert gas P_1 , P_2 , and P_3 , different heat-up rates HR_1 , HR_2 , and HR_3 , and $P_1 < P_2 < P_3$ and $HR_1 < HR_2 < HR_3$.

Cooling of the preform or product after soaking is not particularly critical except as it may result in cracking of the preform or product as the result of the development of undesirable thermal stresses. Cooling rates less than 10°C/min to a temperature of about 100°C are typically used to prevent cracking due to thermal shock. Somewhat higher, but carefully controlled, cooling rates may however, be used to obtain a "sealed skin" on the open cell structure of the product as described below. The rate of cooling below 100°C is in no way critical.

After expanding the coal particulate as just described, the porous coal product is an open celled material. Several techniques have been developed for "sealing" the surface of the open celled structure to improve its adhesive capabilities for further fabrication and assembly of a number of parts. For example, a layer of a commercially available graphitic adhesive can be coated onto the surface and cured at elevated temperature or allowed to cure at room temperature to provide an adherent skin. Alternatively, the expansion operation can be modified by cooling the expanded coal product or preform rapidly, e.g., at a rate of 10°C/min or faster after expansion. It has been discovered that this process modification results in the formation of a more dense skin on the preform or product which presents a closed pore surface to the outside of the preform or product. At these cooling rates, care must be exercised to avoid cracking of the preform or product.

After expanding, the porous coal-based preform or product is readily machineable, sawable and otherwise readily fabricated using conventional fabrication techniques.

5

Subsequent to production of the preform or product as just described, the preform or product may be subjected to carbonization and/or graphitization according to conventional processes to obtain particular properties desirable for specific applications of the type described hereinafter. Ozonation may also be performed, if activation of the coal-based expanded product would be useful in a final product application such as in filtering of air. Additionally, a variety of additives and structural reinforcers may be added to the coal-based preforms or products either before or after expansion to enhance specific mechanical properties such as fracture strain, fracture toughness and impact resistance. For example, particles, whiskers, fibers, plates, etc. of appropriate carbonaceous or ceramic composition can be incorporated into the porous coal-based preform or product to enhance its mechanical properties.

The open celled, coal-based preforms or products of the present invention can additionally be impregnated with, for example, petroleum pitch, epoxy resins or other polymers using a vacuum assisted resin transfer type of process. The incorporation of such additives provides load transfer advantages similar to those

demonstrated in carbon composite materials. In effect a 3-D composite is produced that demonstrates enhanced impact resistance and load transfer properties.

The cooling step in the expansion process results in some relatively minimal shrinkage on the order of less than about 5% and generally in the range of from about 2% to about 3%. This shrinkage must be accounted for in the production of near net shape preforms or final products of specific dimensions and is readily determinable through trial and error with the particular coal starting material being used. The shrinkage may be further minimized by the addition of some inert solid material such as coke particles, ceramic particles, ground waste from the coal expansion process etc. as is common practice in ceramic fabrication.

Carbonization is conventionally performed by heating the preform or product under an appropriate inert gas at a heat-up rate of less than about 5° C per minute to a temperature of between about 800° C and about 1200° C and soaking for about 1 hour or less. Appropriate inert gases are those described above that are tolerant of these high temperatures. The inert atmosphere is supplied at a pressure of from about 0 psi up to a few atmospheres. The carbonization process serves to remove all of the non-carbon elements present in the preform or product such as sulfur, oxygen, hydrogen, etc.

20

Graphitization, commonly involves heating the preform or product either before or after carbonization at heat-up rate of less than about 10° C per minute, preferably from about 1° C to about 5° C per minute, to a temperature of between

about 1700° C and about 3000° C in an atmosphere of helium or argon and soaking for a period of less than about one hour. Again, the inert gas may be supplied at a pressure ranging from about 0 psi up to a few atmospheres.

5 The porous coal-based preforms or products resulting from processing in accordance with the foregoing procedures can be used in a broad variety of product applications, some, but not all, of which will now be broadly described.

10 Perhaps the simplest products that could be fabricated using the coal-based porous preforms or products of the present invention are various lightweight sheet products useful in the construction industry. Such products may involve the lamination of various facing materials to the surface of a planar sheet of the preform material using an appropriate adhesive. For example, a very light and relatively inexpensive wall board would simply have paper laminated to its opposing
15 planar surfaces, while a more sophisticated curtain wall product might have aluminum sheet, polymer or fiber-reinforced polymer sheets or even stainless steel sheet laminated thereto. A wide variety of such products that have lightweight, low cost and adequate strength can easily be envisioned for wallboard, structural wallboard, bulkheads, etc. The materials of the present invention exhibit sound
20 insulation and vibration resistance due to excellent sound and vibration damping properties, good thermal insulating properties (less than about 1 watt per meter K thermal conductivity).

Laminates of these materials may even be used to produce heating element incorporating members, since a graphitized core could serve as an electrical heating element when connected to an appropriate source of electrical energy.

5 Similar surface laminated porous preform core based products could also find use in the transportation industry where lighter and, especially fire retardant walls, bulkheads, containers, etc. are in constant demand. Such products would of course require that the expanded coal-based porous core be carbonized as described hereinabove prior to application of the exterior skins, if fire resistance or
10 retardancy is desired.

Yet another product application for the porous coal products of the present invention is as a replacement for the ceramic foam filters currently applied in the filtering of molten metal such as aluminum for the removal of contaminating
15 particulates also called inclusions. The current ceramic foam materials are relatively expensive and extremely friable. It is easily possible to produce a porous coal-based preform of the type described herein having an appropriate pore size and of the same size and shape as the ceramic foam filter using the above described fabrication process, to serve as a molten metal filter of this type. The cost of such a
20 more robust, i.e., less friable, filter would be considerably less than that of a comparable ceramic foam filter.

Yet other product applications for the materials of the present invention reside in the field of heat exchangers. In this application, the heat transfer properties of a graphitized porous coal-based material can be exploited to produce a heat exchanger capable of extracting heat from or adding heat to a fluid (gas or liquid) flowing through porous coal pores. In this case, the coal-based porous product is joined to an appropriate heat transfer mechanism such as an aluminum skin.

As already alluded to, the coal-based porous preforms and products of the present invention can be produced in any solid geometric shape. Such production is possible using any number of modified conventional processing techniques such as extrusion, injection molding, etc. In each of such instances, the process must, of course, be modified to accommodate the processing characteristics of the starting material coal. For example, in extruding such products, as described below, the coal powder starting material is fed by an auger into an expansion chamber where it is expanded and from which it is extruded while still viscous. Upon exiting the extrusion die, the material is cooled to provide a solid shape of the desired and precalculated dimensions. To improve the efficiency, i.e., cycle time of the process, the input material can be preheated to a temperature below the expansion point, e.g., below about 300° C, fed into the auger chamber where additional heat is imparted to the powder with final heating being achieved just before extrusion through the die.

Similar relatively minor process modifications can be envisioned to fabricate the carbon foams of the present invention in injection molding, casting and other similar conventional material fabrication processes.

5 The following examples will serve to illustrate the practice of the invention.

Examples

Example 1

10 As shown in Figure 2, a layer 10 of comminuted bituminous coal ground to a particle size of about -60 mesh and about 2 inch deep is deposited in mold 12 equipped with a cover 16. Mold 12 is assembled from three individual pieces carbon or tool steel pieces, sides 12A and 12B and bottom 12C, all joined together by bolts 11 and lined with a ceramic glaze or spray applied ceramic lining 13. Cover 16
15 includes a similar interior ceramic lining 15 and is attached to sides 12A and 12B with bolts 17 in the final assembly prior to heating. Gaskets 19 are preferably used to insure a tight fit of cover 16 onto sides 12A and 12B. Cover 16 is optionally equipped with a sintered vent plug 20 to permit purging of the interior of mold 12 with non-oxidizing gas. This configuration, incorporating valve 20 also permits
20 pressurization, if desired to control expansion speed and/ or pore size in the final product as described hereinabove. Nitrogen gas is repeatedly introduced through valve 20 to assure that all oxygen in mold 12 is purged (generally 2-4 such purges have been found satisfactory) and to provide a one atmosphere pressure of nitrogen

inside of mold 12. Mold 12 is then heated at a rate of from about 1 to about 10°C/min up to a temperature of about between about 450 and 600°C and held at this temperature sufficient to devolatalize and sinter the cellular product (generally less than about one hour). This treatment results in the production of an open celled expanded coal product 10A as shown in Figure 3. Mold 12 is then cooled to room temperature at a rate of less than about 10°C/min. to a temperature of 100°C; any remaining pressure is then vented through valve 15 and the sample removed from mold 12 by disassembly of mold 12 by disengagement of bolts 11. Expanded coal product 10A is thereby readily removed from mold 14 and is subsequently sawed to the desired dimensions.

Product 10A has a density of between about 0.4 and about 0.6 g/cm³ and demonstrates a compressive strength on the order of between about 2000 and 6000 psi. Thermal conductivity as determined by the guarded heat flow method is below about 1.0 W/m/K.

Example 2

The application of the process of the present invention in an extrusion process is depicted in Figure 4. As shown in that figure, comminuted bituminous coal 22 of a particle size of about -80 mesh is introduced via hopper 24 into chamber 26 equipped with auger 28 that moves particulate coal 18 through chamber 26 and into expansion chamber 30. Chamber 26 is heated by means of a series of barrel heaters 32, 34 and 36 to impart a temperature of less than about 300° C to

particulate coal 18 as it approaches and enters expansion chamber 26. As is conventional practice in extrusion, chamber 26 is divided into a feed section, a compression section and a metering section each defined roughly by the location of barrel heaters 32, 34 and 36 and imparted by the tapered shape of auger 28.

5 Expansion chamber 30 is maintained under a non-oxidizing atmosphere and at a temperature of about 450° C by means of barrel heater 38. Particulate coal 18 expands within chamber 26 to form expanded coal product 40 and, while still viscous, expanded coal product 40 is extruded through a die 42 to form solid shaped product 44 upon cooling to room temperature. Solid shaped product 44

10 demonstrates properties similar to those obtained from the product described in Example 1.

At the point where particulate coal 22 exits chamber 26 and enters expansion chamber 30, chamber 26 is preferably equipped with a breaker plate 46 that serves

15 to break up any large agglomerates of particulate coal 22 that may have formed in transit within chamber 26.

Cellular coal-based extrudate 44 may have virtually any solid shape ranging from a large flat panel 4'X 8' as might be used as the core of the above-described

20 building panel to square shapes, rounds, channels and even tubular shapes if a bridge die is used in the extrusion process. Almost any shape that can be achieved with plastic or metal extrusion can be similarly obtained using the process of the present invention.

As the invention has been described, it will be apparent to those skilled in the
5 art that the same may be varied in many ways without departing from the spirit and
scope of the invention. Any and all such modifications are intended to be included
within the scope of the appended claims.

What is claimed is:

- 1) A semi-crystalline, largely isotropic, porous coal-based product produced from particulate coal of a small diameter, having a density of between about 0.1 and about 0.8 g/cm₃ and a thermal conductivity below about 1 W/m/°K.
- 2) The porous coal-based product of claim 1 having a compressive strength below about 6000 psi.
- 3) The porous coal-based product of claim 1 that has been carbonized.
- 4) The porous coal-based product of claim 1 that has been graphitized.
- 5) A method for producing a porous coal-based product from coal comprising:
 - A) comminuting coal to a small particle size to form a ground coal;
 - B) placing said ground coal in a mold;
 - C) heating said ground coal in said mold under a non-oxidizing atmosphere to a temperature of between about 300° C and about 700° C and soaking at this temperature for a period of from about 10 minutes to about 12 hours to form a preform; and

D) controllably cooling said preform.

6) The method of claim 5 wherein said inert atmosphere is applied at a pressure of from about 0 psi up to about 500 psi.

5

7) The method of claim 5 wherein said temperature is achieved using a heat-up rate of between about 1° C to about 20° C per minute.

10

8) The method of claim 5 wherein said controlled cooling is accomplished at a rate of less than about 10° C/min to a temperature of about 100° C.

9) A laminated sheet comprising:

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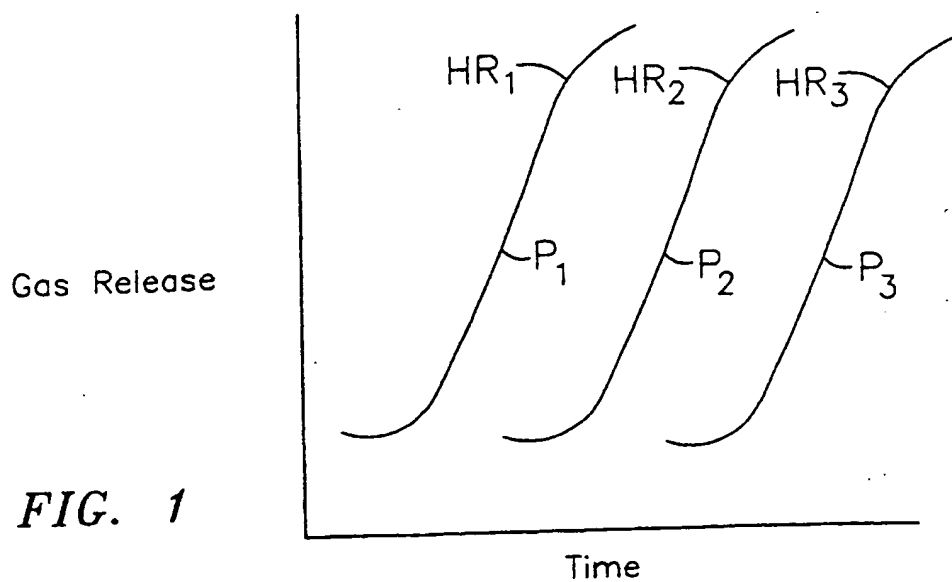
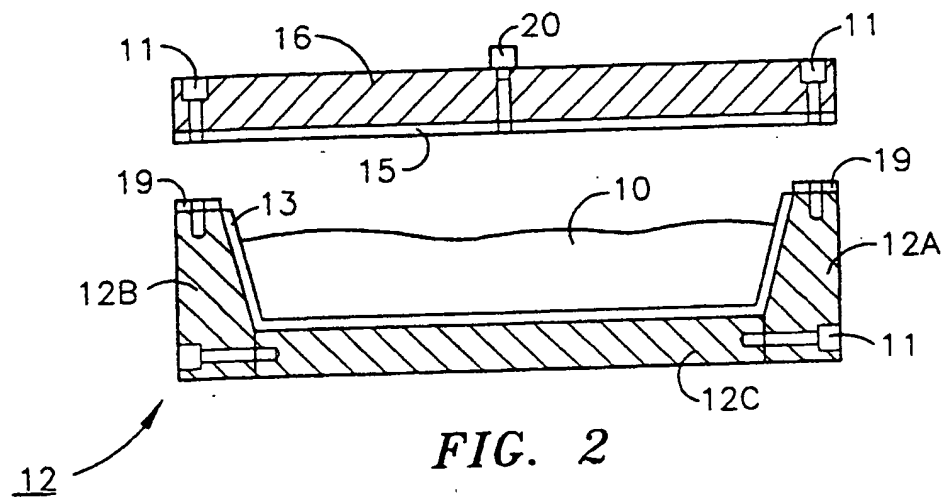
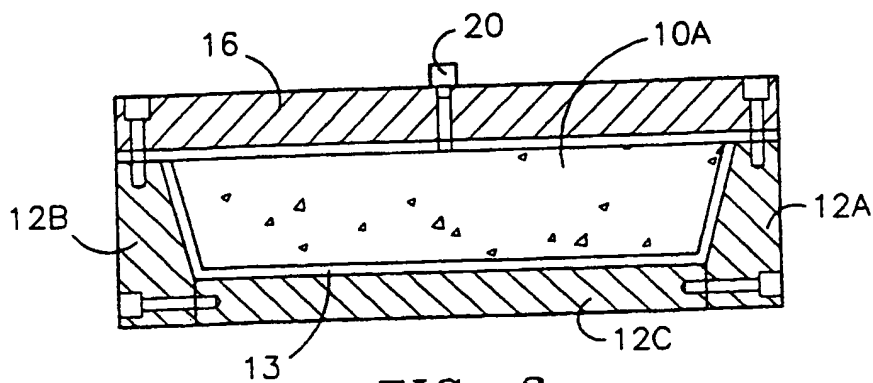
- A) a pair of skins laminated to either side of;**
- B) a core of a semi-crystalline, largely isotropic, porous coal based product produced from particulate coal of a small diameter, having a density of between about 0.1 and about 0.8 g/cm³ and a thermal conductivity below about 1 W/m/°K.**

20

10) The laminated sheet product of claim 9 wherein said skins comprise a material selected from the group consisting of aluminum, steel,

polymer sheet, inconel, titanium, refractory metals, fiber reinforced
polymer sheet and paper.

- 5
- 11) The laminated sheet product of claim 9 wherein said sheet core has
been carbonized.
 - 12) The laminated sheet product of claim 9 wherein said sheet core is
graphitized.

**FIG. 1****FIG. 2****FIG. 3**

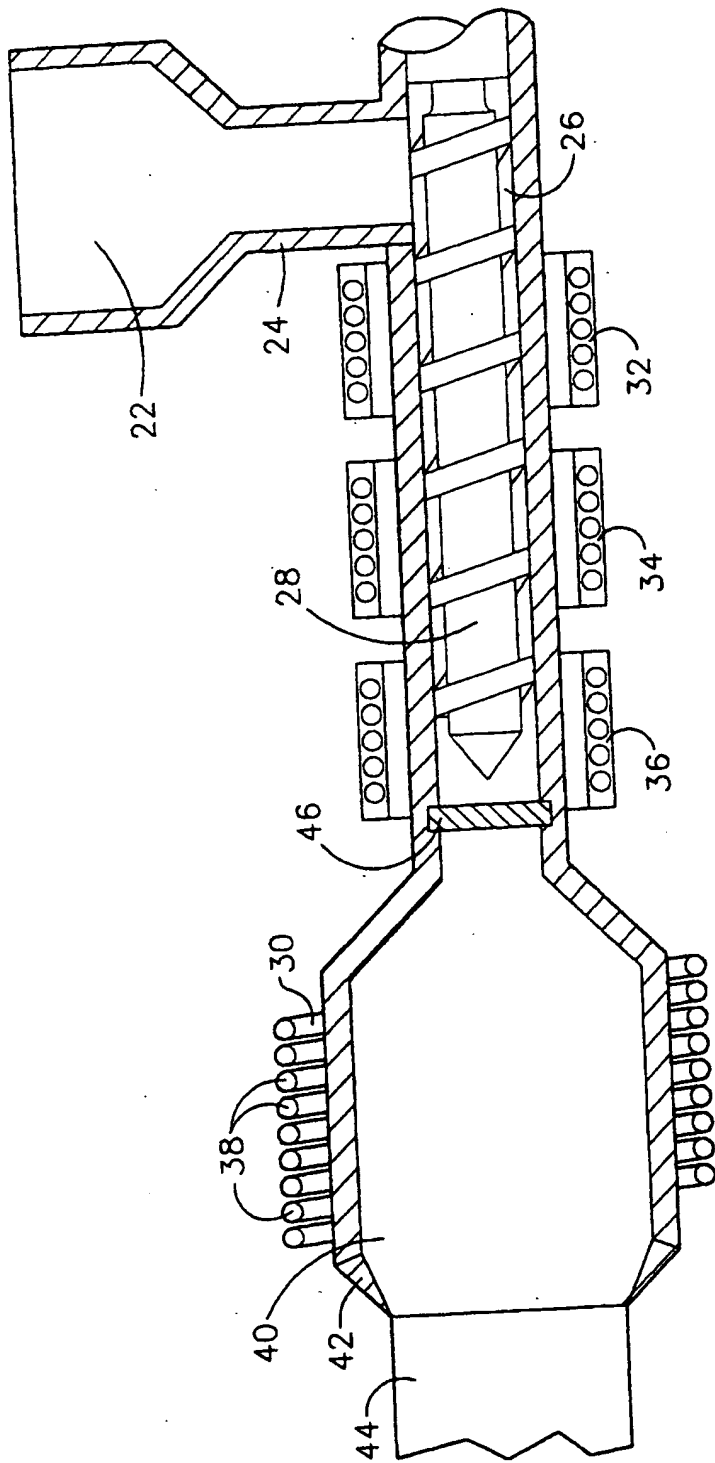


FIG. 4

INTERNATIONAL SEARCH REPORT

International application No.
PCT/US00/32630.

A. CLASSIFICATION OF SUBJECT MATTER IPC(7) : C10L 9/08; B32B 5/16 US CL : 44/591, 607, 620; 201/6; 264/29 According to International Patent Classification (IPC) or to both national classification and IPC		
B. FIELDS SEARCHED Minimum documentation searched (classification system followed by classification symbols) U.S. : 44/591, 607, 620; 201/6; 264/29 Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched NONE Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) NONE		
C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
X	US 3,309,437 A (HARNETT) 14 March 1967, note in entirety, Table 1, Examples 4 and 5; col. 5, lines 20-44; Table 1, Examples	1-4
—		—
Y	1-12, col. 1, lines 51-col. 2, lines 1-35, col. 3, lines 1-22 and 49 to col. 4, lines 1-65 and col. 6	5-12
Y	US 4,127,391 A (KOPPELMAN) 28 November 1978, cols. 6 and 7, bridging paragraph.	1-12
Y	US 4,917,923 A (YOSHIDA et al) 17 April 1990, note col. 4, lines 29-end; col. 7, lines 26-48 and col. 9	1-12
Y	JP 08112876 A (KURODA S) 07 May 1996, note English abstract and fig. 1.	1-12
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INTERNATIONAL SEARCH REPORT

International application No.
PCT/US00/32630

C (Continuation). DOCUMENTS CONSIDERED TO BE RELEVANT

Category*	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No.
A	US 3,018,227 A (BAUM et al.) 23 January 1962, note fig. 1 and claims 1-11.	1-12
A	US 3,185,635 A (CREGLOW) 25 May 1965, note fig. 7 and claims 1-27.	1-12
A	US 3,754,876 A (PENNINGTON et al.) 28 August 1973, see abstract, fig. 1 and claims 1-12.	1-12
A	US 4,632,731 A (BODLE et al.) 30 December 1986, see abstract, fig. 1 and claims.	1-12

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